Seat	No.:	
Dout	110	

Enrolment No.____

GUJARAT TECHNOLOGICAL UNIVERSITY

BPHARM – SEMESTER II • EXAMINATION – SUMMER • 2014

Subject code: 220006 Date: 06-06-2014

Subject Name: Physical Pharmacy

Time: 02:30 pm - 05:30 pm Total Marks: 80

Instructions:

- 1. Attempt any five questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a) (b) (c)	Classify complexes and write on metal complexes in detail. Differentiate between channel lattice type and clathrates complexes. Enumerate methods of analysis for complexes. Write on pH titration method.		
Q.2	(a)	Write in brief on any two:	06	
	(b) (c)	i) Protein binding ii) Pharmaceutical application of colloids iii) Association colloids What are kinetic properties of colloids?Explain electrical properties of colloids.	05 05	
Q.3	(a)	Write on:	06	
	(b) (c)	i)Particle and particle size distribution ii)Number and weight distributionExplain derived properties of powders.Give in brief methods to determine particle shape and surface area.	05 05	
Q.4	(a)	Give Newton's law of flow. Define kinematic viscosity. Give relation between	06	
	(b) (c)	temperature effect on viscosity of liquids. Write briefly on any one: i) Plastic Flow ii) Pseudoplastic Flow iii) Dilatant Flow Write on Thixotropy, measurement of thixotropy and its applications in pharmaceutical formulations.	05 05	
Q.5	(a)	 Write in short on any one: i) Sedimentation of flocculated particles ii) Difference between flocculated and deflocculated suspensions. iii) Formulation of suspensions. 	06	
	(b)	Give types, applications and preservation of emulsions.	05	
	(c)	Write on Rheological properties of emulsions.	05	
Q. 6	(a)	Write in brief on (any two): 1. Spreading Coefficient 2. CMC 3. Surface Active Agents 4. Systems of HLB classification	06	
	(b)	Write on: Surface and interfacial tensions and techniques of their measurement. Write	05	
	(c)	in brief on Surface Free Energy. Write in details on: Physical Stability of emulsions.	05	
	(C)			
Q.7	(a)	Explain Liquefaction of gases and Methods of achieving liquefaction. Give application of this phenomenon in Pharmacy.	06	
	(b)	Explain the following(any one):	05	
		i) Crystalline solids ii) Polymorphisms iii) Amorphous solids	^=	
	(c)	Discuss 2 component systems containing liquid phases or Ternery systems with one pair of partially miscible liquids.	05	
