GUJARAT TECHNOLOGICAL UNIVERSITY

B. Pharm. - SEMESTER II • EXAMINATION - WINTER 2015

Subject Code: 220006 Date: 21/12/2015 **Subject Name: Physical Pharmacy**

Time: 2.30 pm to 5.30 pm **Total Marks: 80**

Instructions:

- 1. Attempt any five questions.
- 2. Make suitable assumptions wherever necessary.

in solubility of liquids in liquids.

(b)

suspension.

- 3. Figures to the right indicate full marks.
- **Q.1** What is thixotropy and antithixotropy? Draw different types of thixotropic and **06** antithixotropy curves and explain the mechanism for their behavior with suitable examples. Write a short note on Gel-Sol-Gel Transformation. 05 **(b)** Give the working principle of cup and bob viscometer with a labelled diagram. 05 (c) 06 **Q.2** Discuss the capillary rise method for measuring surface tension. (a) Write a short note on Spreading Coefficient. 05 **(b)** Discuss the electric properties of interfaces. **(c)** 05 0.3 Explain the particle volume measurement by coulter counter method. **06** (a) Explain derived properties of powders. **(b)** 05 Define Reynolds number. Write Stokes' equation. What are the applications in 05 (c) pharmacy? **Q.4** What are association colloids? Mention the mechanism of formation of 06 micelles. Write a note on Controlled Flocculation. **(b)** 05 Write a note on Hydrophilic Lipophilic Balance. 05 (c) 0.5 State the phase rule. Explain Phase equilibria for system containing two 06 (a) components. Write a note on polymorphism. Describe its importance with example. 05 **(b)** (c) What are kinetic properties of colloids? 05 **O.** 6 (a) Discuss the effect of pressure, temperature and chemical reaction for solubility 06 of gasses in liquids. Explain Buffer Equation. What is buffer capacity? How it is calculated. 05 **(b)** Write a note on physical stability of emulsion. 05 (c) **Q.7** Differentiate ideal and real solution. Explain the influence of foreign substances (a) 06

Define suspension and differentiate between flocculated & deflocculated

Enlist the methods for adjusting tonicity and pH. Explain any one.

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