Seat No.:	Enrolment No
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GUJARAT TECHNOLOGICAL UNIVERSITY B.PHARM – SEMESTER – 3- EXAMINATION –WINTER - 2018

Subject Code:230001			
Tim	e:10: uction 1. 2.	Attempt any five questions. Make Suitable assumptions wherever necessary.	
	3.	Figures to the right indicate full marks.	
Q.1	(a) (b) (c)	State Raoult's law. Explain positive and negative deviations from Raoult's law with suitable examples. Define: Molarity, Normality and Molality. State the Henry's low. Describe Ostwald-Walker method of measuring the relative lowering of vapour pressure.	06 05 05
Q.2	(a) (b) (c)	Describe Arrhenius theory of electrolytic dissociation. Describe the Beckmann method to determine freezing point lowering. State & Explain Faraday`s laws of Electrolysis.	06 05 05
Q.3	(a) (b) (c)	Discuss various factors affecting on rate of reaction. Explain various methods to determine the order of reactions. Define first order reaction. Derive equations for first order reaction and its half-life and self-life.	06 05 05
Q.4	(a) (b) (c)	Define Complexation. Classify various types of complexes. Write applications of complexes in pharmacy. Write note on protein binding.	06 05 05
Q.5	(a) (b) (c)	Define and Classify the polymers with examples. Write a note on: Characterization of polymers. Describe the pharmaceutical applications of polymers.	06 05 05
Q. 6	(a) (b) (c)	Define Diffusion and Flux. Explain Fick's first and second law of diffusion. Define Dissolution. Explain the Noyes Whitney equation for describing rate of drug dissolution. Describe the USP type-I & II dissolution apparatus with labeled diagram.	06 05 05
Q.7	(a) (b) (c)	Write a short note on Accelerated stability study. Describe Debye-Heckel theory for strong electrolytes. Discuss the general properties of polymer solutions.	06 05 05
