

GUJARAT TECHNOLOGICAL UNIVERSITY
B.Pharm – SEMESTER V • EXAMINATION – Summer 2017

Subject Code:2250003

Date: 06/06/2017

Subject Name: Pharmaceutical Analysis III

Time: 10.30 AM to 01.30 PM

Total Marks: 80

Instructions:

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

Q.1	(a) Define chemical shift. Explain factors affecting chemical shift.	06
	(b) Describe instrumentation of PMR.	05
	(c) Write a brief overview of C13 NMR spectroscopy.	05
Q.2	(a) Explain: base peak, metastable ion peak and isotopic ion peak.	06
	(b) Discuss principle of Mass Spectroscopy.	05
	(c) Explain Mc-Lafferty rearrangement.	05
Q.3	(a) Write a note on FT-IR.	06
	(b) Discuss radiation sources used in IR spectroscopy.	05
	(c) Describe sample handling techniques used in IR spectroscopy.	05
Q.4	(a) Write the difference between AAS and AES.	06
	(b) Describe the principle of flame photometer.	05
	(c) Write a note on hollow cathode lamp.	05
Q.5	(a) Explain Auxochrome, molar absorptivity and isosbetic point.	06
	(b) What is EMR? Explain wave properties of EMR	05
	(c) Describe principle of UV spectroscopy.	05
Q. 6	(a) Write the difference between fluorescence and phosphorescence.	06
	(b) Draw a well-labeled diagram of spectrofluorimeter.	05
	(c) Write a brief over view of application of UV-visible spectroscopy.	05
Q.7	(a) Describe calibration of UV-visible spectrophotometer as per IP.	06
	(b) Give the fragmentation pattern of the following compounds	05
	(i) Benzyl alcohol	
	(ii) Benzamide	
	(c) How will you differentiate between the following pairs of compounds using IR Spectroscopy	05
	(i) $\text{CH}_3\text{CH}_2\text{NH}_2$ and $\text{CH}_3\text{CH}_2\text{OH}$	
	(ii) CH_3COOH and CH_3CHO	
