Seat No.:	Enrolment No.

GUJARAT TECHNOLOGICAL UNIVERSITY B.PHARM – SEMESTER – 7- EXAMINATION –WINTER - 2018

Subject Code: 270004 Date: 22/11/2018

Subject Name: Pharmaceutical Analysis III

Time: 10:30 AM TO 01:30 PM Total Marks: 80

Instructions:

1. Attempt any five questions.

- 2. Make Suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a)	Define: a) Line Spectra b) Spectroscopy c) Emission Spectroscopy	06
	(b)	Enlist IR detectors. And Explain any one in detail.	05
	(c)	Write a note on interference in Atomic Absorption Spectroscopy.	05
Q.2	(a)	Explain principle and working for Time of Flight mass Analyser.	06
	(b)	Explain the different types of vibrations in IR.	05
	(c)	Write a note on Hollow Cathode Lamp.	
Q.3	(a)	Explain the principle of NMR.	06
	(b)	Define and derive Beer Lambert's Law.	05
	(c)	Explain the principle of Florescence and Phosphoresce with Jablonski diagram.	05
Q.4	(a)	Explain: a) Bathochromic Shift b) Auxochrome c) Hypsochromic shift	06
	(b)	Explain: a) Mc lafferty Rearrangement b) Nitrogen Rule	05
	(c)	Write a note on Deviations of Beer Lambert's Law.	05
Q.5	(a)	Define Quenching. Write a note on Quenching Agents.	06
	(b) Enlist Ionization techniques for Mass Spectroscopy and explain any one		05
	(c)	Write a note on chemical shift for NMR.	05
Q. 6 (a)		Describe the instrumentation required for fluorescence analysis. What are	06
		primary filter and secondary filters?	
	(b)	Write a note on Radiation sources for UV spectroscopy.	05
	(c)	Explain: a) Fingerprint region b) Base ion peak	05
		Comment: Benzene is colourless while Nitrobenzene is pale yellow in colour.	06
		Which elements are contributed to M+2 ion peak in mass spectra? Why?	05
	(c)	Write a note on FT-TR.	05
